Carbon Capture & Storage – CO₂ mineralisation via fuel cells

innovative chemistry to generate zerocarbon electricity & construction materials at global scale

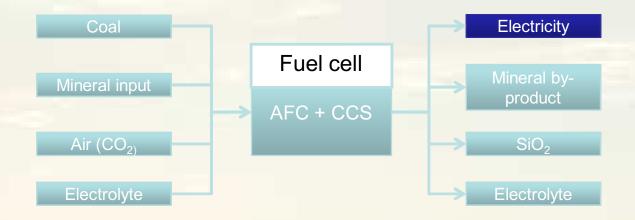
Cleanpower 2010 (http://bit.ly/cleanpower)
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Summary

- Cambridge Carbon Capture Who we are
- Conventional CCGS economic challenge
- CCS mineralisation What, How & Why
- CCC's unique chemistry using fuel cells
- Markets for mineralisation

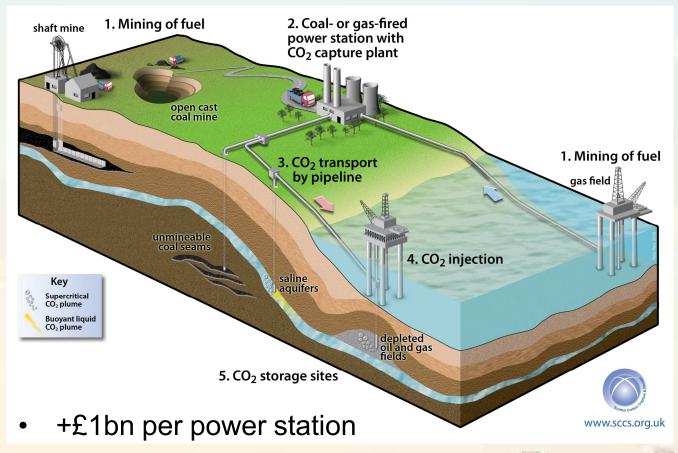
Cambridge Carbon Capture

- new 2010 venture
- developer of innovative CCS chemistry to make CCS profitable
- unique IP: fuel cells (CO₂ capture) + mineralisation (CO₂ sequestration)



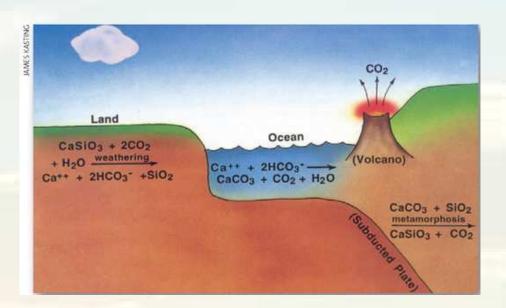
- located at Hauser Forum & Dept. Materials
- Team: Department of Materials; Judge Business School; Nottingham CICCS; Cambridge Enterprise; TSB; EEDA; Renewables East

CCGS – economically unfeasible?



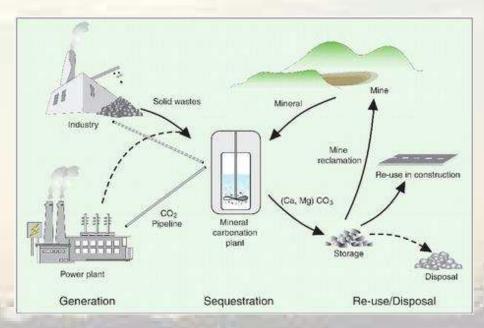
- •Will CCS costs fall fast enough?
- •When will ETS price > costs?
- Decades untilCCS investable?
- •CO₂ storage fees?
- •Will the public accept CO₂?
- •Utilities have other options.
- •China will not let CCS risk growth.
- ~30% parasitic energy; ~30% electricity cost increase
- £/MWh CCS coal > £MWh nuclear & onshore wind
- ~€40-90/tonne CCS >> €15/tonne ETS price

CCS Mineralisation – what is it?



- CCS Mineralisation is a very fast version of natural mountain weathering & marine sedimentation processes
 - CO₂ + H₂O + Ca/Mg silicates = limestone
 - >99% world's carbon reservoir is locked in limestone & dolomite

- Known deposits serpentine & olivine > amount needed to sequester known coal reserves
- Challenge is to develop a practicable industrial process
 - reaction kinetics
 - capture ratio (tonnes CO₂ per tonne feedstock)
 - energy & cost
 - materials logistics



CCS Mineralisation – how to do it?

- IPCC & IEA review reports examine range of chemistry options & costs (+10yrs of research):
 - many process papers & patents; little data on energy consumption
 - milling to <75um, heating ~650C and/or acid/base digestion (~100C) required to activate serpentine for carbonation
 - aqueous phase chemistries better than gas-solid; serpentine carbonation at ~80-150C & ambient to high-pressure
 - watching brief with scope for new, cheaper chemical processes
 - continuing progress in low-energy pre-treatment of serpentine (olivine, wollastonite & industrial wastes also offer low-energy routes)
- wide range of feedstock mineral/waste options any Ca or Mg ions
- e.g. closed-cycle, pH-swing ammonium bisulphate digestion at 80C & carbonation to convert asbestos (Mg₃Si₂O₅(OH)₄) waste to high-purity MgCO₃, SiO₂ & Fe
- e.g. mine tailings study at four Canadian & Australian sites: 1-50 kt/CO₂/yr per mine site (CO₂ from atmosphere) rate-limited by silicate mineral dissolution & depends on local climate [Dipple, 2009]

CCS Mineralisation – why do it?

- huge capacity (of feedstock minerals) globally distributed >> conventional CCS
- option where no suitable underground oil/gas/aquifer space exists
- geologically stable CO₂ storage as limestone avoids environmental, safety & legacy concerns of conventional CCS i.e. higher public acceptability v. "toxic" CO₂
- higher density CO2 storage (1.6tonnes CO₂/m³ for MgCO₃) >2X supercritical CO₂
- endothermic process, in principle (not yet in practice); faster, lower-energy chemistries available
- materials product value makes mineralisation cheaper than conventional CCS
 - JBS economic modelling "CCC best case" electricity is cheaper than unabated coal
- sequestered carbonate matches scale of market for aggregates & fillers in global construction industry
 - global ~5bn tonnes/year CO₂ from coal-fired power generation
 - 100% mineralisation would generate ~20bn tonnes/yr solid carbonates
 - global ~25bn tonnes construction aggregates (of which cement ~3bn tonnes)
- potential by-product materials & process values:
 - recovery of residual metals (Fe, Co, Ni, rare-earths, precious metals)
 - high-purity silica, bicarbonate chemicals
 - cementitious phase carbonates to substitute Portland & pozzolanic cements
 - remediation value from carbonation of landfill wastes, mine tailings & hazardous wastes
 - CO₂ credits from displaced quarrying & cement processes

CCS Mineralisation – Why isn't industry doing it?

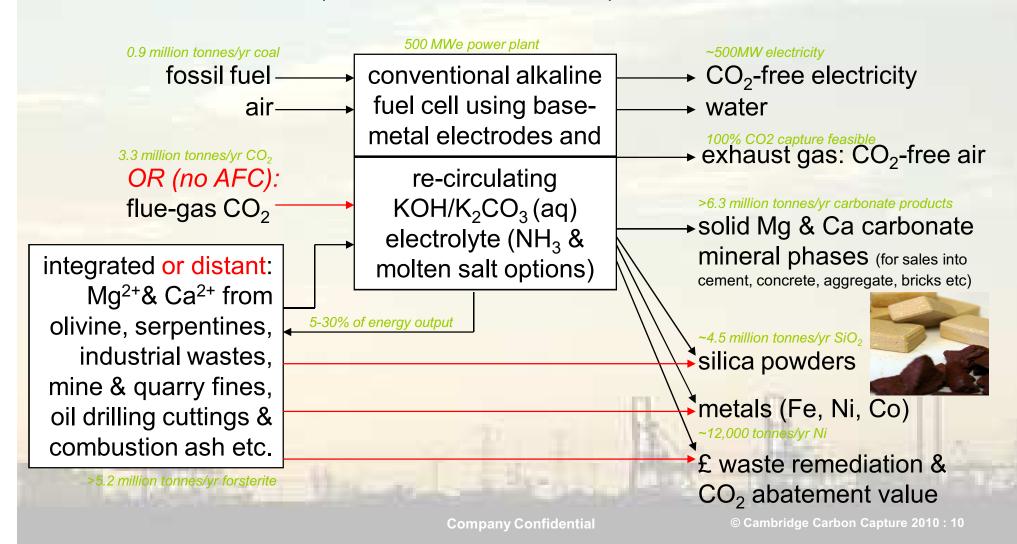
- CCS industry has so-far largely dismissed mineralisation on basis that is more energy intensive, costly & logistically difficult than conventional CCGS:
 - cost & energy of intense natural mineral pre-treatments (Mg-extraction)
 - still have energy & capital costs of CO₂ capture (most expensive part of CCS)
 - millions tonnes of mineral feedstocks to be mined & transported & process inputs
 - millions tonnes of waste carbonate or bicarbonate solids to store or dispose of
 - overall cost ~€30-100/tonne v. €30-45/t in 2030 (€60-90/t in 2015) for CCGS
- CCS industry is focussed on government-funded delivery of power-stationscale demonstrations of EOR-proven CO₂ separation, compression & underground storage
- But... industries, governments, researchers, companies & investors are increasingly looking at feasibility & re-evaluating economics:
 - ETI £1m 2011 paper study launched on "mineralisation opportunities" (Shell, Caterpillar, BGS, CICCS) – UK mineral distribution + techno-economic viability
 - Australia Latrobe valley; NSW (GreenMag-U.Newcastle)
 - USDoE CO2 Mineral Seqstr. WG (ARU, ASU, LANL, NETL, PSU, SAIC, UU)
 - CCC, Calera, ICS, Novacem, Carbon8, Calix, Oxford Geo-Engineering, Skyonic

CCC's Unique Technology

- Oxidation of carbon to solid carbonate (~470kJ/mole) releases ~15% more energy than oxidation (~400kJ/mole) to CO₂
- (hydro)carbons can be efficiently, cleanly & cheaply converted to electricity via direct electro-chemical oxidation in an alkaline fuel cell
 - at high efficiency >50% (+ le Chatelier) & with low-cost base-metal electrodes
 - currently using methanol (for proof of principle); low-rank coals feasible
- Aqueous KOH/K2CO3/KHCO3 & ammonia-based solutions
 - CCS-proven fast capture & conversion medium for CO₂ to (bi)carbonate;
 - is an excellent electrolyte & existing component of alkaline fuel cell systems
 - after carbonation, can be regenerated by reaction with Mg²⁺ & Ca²⁺ ions from abundant silicate minerals and alkaline industrial wastes & slags.
 - OH- and ammonium salt digestion reported for serpentine
- Process chemistry can also be used for point-source CO2 sequestration
- Carbonate mineral precipitates can be consolidated to building materials & aggregates and cementitious phases can replace high value cements & displace additional CO₂

CCC Process

$$C + O_2 + 2OH^- \xrightarrow{KOH} CO_3^{2-} + H_2O \xrightarrow{Ca^{2+}, Mg^{2+}} CaCO_3 + 2OH^- dG = -ve$$



Markets for CCC's Process

- £330 billion/yr global market for carbonate materials produced by CCC process
- £3300 bn (period 2010-2050) global spend on CCS technology (IEA estimate)
- £300 billion/yr ~7.5TWh coal-fired electricity @ £40/MWh wholesale
- £130 billion/yr (~10bn tonnes CO₂ @ £13/tonne)
- global power growth ~2.5% + ~2% replacement
- Early-stage opportunities
 - Minerals & mining industry: recovery of high-value metals & chemicals; on-site clean electricity;
 - Industrial waste processing industry: stabilisation process profitable via avoidance of landfill taxes
 - Industrial AFC industry: improve the business case for fuel cells
- Later-stage opportunities
 - Utilities: £/MWh coal+MinCCS < £/MWh coal+€15/t
 - Cement & concrete industry: low-CO₂ manufacturing process
 - Construction sector: low-embodied carbon building materials
 - CO2 credits & CDM (credit for saving CO2 overseas)
 - optimisation of mineralisation chemistry to maximise value of materials throughput
 - continuous improvement in serpentine activation processes driven by market demand

Conclusions

- Mineralisation offers a highly profitable & fully scalable approach to CCS for a ~\$1 trillion/year market
- Technology is at least as proven as CCGS
- Commercial deployment doesn't depend on learning curves, carbon pricing or CO₂ pipeline infrastructure
- Can start small-scale and build from there
- …and it's only just coming onto the CCS radar screen

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